

7231

BOARD DIPLOMA EXAMINATION, (C-20) OCTOBER/NOVEMBER—2023

DCHST - THIRD SEMESTER EXAMINATION

PHYSICAL AND ORGANIC CHEMISTRY

Time: 3 Hours [Total Marks: 80

PART—A

 $3 \times 10 = 30$

Instructions: (1) Answer **all** questions.

- (2) Each question carries three marks.
- (3) Answers should be brief and straight to the point and shall not exceed five simple sentences.
- **1.** Define heat of neutralization. Give one example.
- **2.** Write any three characteristics of chemical equilibrium.
- **3.** Explain homogeneous equilibrium with one example.
- **4.** State any three differences between lyophilic and lyophobic colloids.
- **5.** Define (a) emulsification and (b) emulsifier.
- **6.** Explain ethyl chloride preparation from Grove's process.
- **7.** Explain the properties of ethyl alcohol reaction with (a) acetic acid and (b) conc. H₂SO₄.
- **8.** Explain sp hybridization.
- **9.** Explain the unique character of carbon.
- **10.** Write any three uses of toluene.

PART—B 8×5=40

Instructions: (1) Answer **all** questions.

- (2) Each question carries eight marks.
- (3) Answers should be comprehensive and criterion for valuation is the content but not the length of the answer.
- **11.** (a) Explain first law of thermodynamics and derive its mathematical equation.

(OR)

- (b) Explain endothermic and exothermic reactions with examples.
- **12.** (a) Explain Le Chatelier's principle for the preparation of ammonia by Haber's process.

$$N_2(g) + 3H_2(g) \rightleftharpoons 2NH_3(g)$$
(OR)

- (b) Explain the effect of temperature, pressure and concentration on equilibrium.
- **13.** (a) Explain the methods to express concentration of solutions.

(OR)

- (b) Write the preparation methods of colloids.
- **14.** (a) Explain the properties of ethyl chloride reaction with (i) KCN and (ii) AgCN.

(OR)

- (b) Explain two preparation methods and two chemical properties of alcohols.
- **15.** (a) Explain two preparation methods and two chemical properties of toluene.

(OR)

(b) Explain two preparation methods and two chemical properties of phenol.

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PART—C 10×1=10

Instructions: (1) Answer the following question.

- (2) The question carries **ten** marks.
- (3) Answer should be comprehensive and the criterion for valuation is the content but not the length of the answer.
- **16.** Explain any two preparation methods of chloroform and write any two chemical properties of it.

