

$\label{eq:cost-alpha} \text{C09-A/AA/AEI/C/CM/EC/EE/} \\ \text{CHST/FW/IT/M/MNG/MET/PKG/TT-} \\ \textbf{104}$

3004

BOARD DIPLOMA EXAMINATION, (C-09) OCTOBER/NOVEMBER-2018 FIRST YEAR EXAMINATION

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 Hours] [Total Marks: 80

PART-A

3X10=30

Instructions:

- 1. Answer **All** questions.
- 2. Each question carries **Three** marks.
- 3. Answer should be brief and straight to the point and shall not exceed five simple sentences.
- 1. Write a short note on metallic bond.
- 2. Define oxidation. Give an example.
- 3. Define Normality. Write the formula for Normality.
- 4. Discuss the Strength of acids and bases according to Arrhenius acid base theory.
- 5. What are strong electrolytes and weak electrolytes? Give an example for each.
- 6. State three disadvantages of using hard water in industries.
- 7. List out any three advantages of plastics over traditional materials.
- 8. Define primary and secondary fuels giving one example each.
- 9. Write a brief note on Green House effect.
- 10. Define the following terms with examples (i) Decomposers (ii) Consumers.

PART-B

10X5=50

Instructions:

- 1. Answer any **Five** questions
- 2. Each question carries **ten** marks.
- 3. Answer should be comprehensive and the criterion for valuation is the content but not the length of the answer
- 11. What are quantum numbers? Explain the significance of various quantum numbers.
- 12. (a) Define terms (i) mineral (ii) Ore (iii) Gangue (iv) Flux (v) Slag.
 - (b) Explain the concentration of ore by forth floatation process.
- 13. (a) Define Galvanic Cell. Explain the construction of galvanic cell with an example.
 - (b) Write any four differences between electrolyte cell galvanic cell.
- 14. (a) Describe the termination of (i) composition cell (ii) Stress cell (iii) Concentration cell.
 - (b) Explain the different types of protective coatings used in prevention of corrosion.
- 15. (a) Explain the municipal method of treatment of water for drinking purpose with a net diagram.
 - (b) What is temporary hardness and permanent hardness? Write its causes.
- 16. (a) Explain the processing of Natural rubber from latex.
 - (b) Give a method of preparation and two uses of (i) Polythene and (ii) PVC.
- 17. (a) What are the effects of deforestation?
 - (b) Explain any three methods of control of air pollution.
- 18. (a) Define molarity calculate the molarity of 2 litre solution containing 4 gm of NaOH (GMW of NaOH=40)
 - (b) Define buffer solution. What are different types of buffer solutions? Give an example for each type.
